Comparison between TiO$_2$/rGO and TiO$_2$ with Various Microelectrode Gaps for Biosensor Applications

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ABSTRACT

Sensitive, selective, and stable materials are important to act as an electron transport layer to construct signals in biosensors. Graphene-based biosensors get attention in this industry due to their selective and sensitive properties. We fabricated TiO$_2$/rGO and TiO$_2$ using spin coating in this study. Thermal evaporation deposited the silver. The picoammeter examined the influence of microelectrode gap sizes (50, 100, and 250 μm) on electrical performance. It was revealed that a smaller gap generates better electrical performance. At the smallest gap, the 50μm TiO$_2$ (8.04Å-12 Å) and TiO$_2$/rGO (2.385Å-8 Å) at 0.04V. The morphological and chemical analysis supported the electrical behavior of TiO$_2$/rGO and TiO$_2$. These results show that the method used was able to form a TiO$_2$/rGO film that is sensitive to the size of the gap between two microelectrodes.

Keywords: Source-drain Gap, Sol-gel method, TiO$_2$, TiO$_2$-rGO

1. INTRODUCTION

A biosensor is an analytical device that detects the presence of a chemical compound by utilising a biological component with a physicochemical detector. Since the 1960s, many sensors for biosamples have been discovered. For example, piezoelectric biosensors [1], [2] enzyme-based biosensors [3], and deoxyribonucleic acid (DNA) biosensors [4]. A transducer, such as an electrical signal [5], translates the response from a biological sample into a measurable signal [5]. The choice of transducer for each sensor depends on the mechanism by which the sensors work. Generally, transducer materials are classified as conductors, insulators, or semiconductors [2]. Metal oxide semiconductors, such as titanium dioxide (TiO$_2$), tin oxide (SnO$_2$), and zinc oxide (ZnO), are commonly used due to their large surface area, biocompatibility, high stability, and low cost. The biosensors industry widely employs metal oxide semiconductors [3]–[5]. One of the materials used in this research is titanium dioxide, as a highlight.

TiO$_2$ is a non-toxic material with uncommon chemical and physical properties. Typically, TiO$_2$ is an n-type semiconductor with an optical band gap of 3.2 eV. Because of its comparatively good performance and stability, TiO$_2$ is commonly used in the fabrication of electronic and optoelectronic devices, for example, photodetectors [6], and dye-sensitised solar cells (DSSCs) [7]. A few studies show that TiO$_2$-based sensors have high selectivity and sensitivity for detecting biomolecules such as glucose [8].

Other than semiconductors, graphene is also a common material in the electrical industry. Graphene is a single layer of carbon atoms tightly arranged in a honeycomb configuration and exhibiting high charge carrier mobility, high conductivity, and mechanical strength. It can be synthesised in several methods such as mechanical exfoliation [9], epitaxial growth [10], ion-beam irradiation [11], and chemical vapour deposition (CVD). It is the most favourable transducer material for use in sensors [3]. However, the absence of defects in pure graphene leads to difficulties in functionalization. Other derivatives of graphene, such as graphene oxide (GO) and reduced graphene oxide (rGO), have the potential for bioreceptor immobilisation due to the abundance of functional groups in their structures [12].

At the moment, there is a lot of excitement in the industry about carbon-based sensors, such as GO-based sensors [13]. Due to its unique properties, 2D material has high electron mobility, high electrical conductivity, and high thermal conductivity. This is because oxygenated functional groups pass through the edges and planes of the layered sp3 carbon atoms [14]. The GO-based biosensor served as a flexible biosensor for wearable devices. Chemically oxidizing graphene to form graphene oxide led to the discovery of improvements in graphene as an electron transport layer (ETL) [15]. It shows better physical and electrical properties or carbon-based biosensors [12].

Modification of transducer material revealed an improvement in the selectivity and sensitivity of biosensors rather than single-material transducers. As an illustration, previous research revealed that both TiO$_2$–graphene biosensors and TiO$_2$ biosensors can detect glucose at 6.2 mM and 6.7 mM [8], [16]. This is due to the combination of unique properties of TiO$_2$ and graphene.
that is able to enhance the synergistic effect between glucose and the transducer [16]. Furthermore, the composition of TiO₂ and graphene creates a wider surface-to-volume ratio that facilitate electron transfer between biosample and electrode through electron injection to graphene from the TiO₂ conduction band [17], [18]. Every modified or composite material has an attractive property.

We investigated the effect of fabricating two microelectrodes on TiO₂ and TiO₂/rGO with different gap sizes (50, 100, and 250µm). We also report on the performance of both materials which, have excellent electrical properties.

2. THEORETICAL BACKGROUND

The working principle of biosensors is most likely similar to that of a metal oxide field effect transistor (MOSFET). The fabrication of FET commonly involves photolithography, etching, [19], [20] and metallization. The transducer material in the channel region acts as a connection, converting the external charge of biomolecule interaction. Previous work [21], has explained further information on the theoretical algorithm.

3. METHODOLOGY

3.1. Synthesis of TiO₂/rGO Thin Film

There are a few methods used to synthesise TiO₂ solutions, such as the hydrothermal method, the sol-gel method, and wet-chemical synthesising by precipitation of hydroxides from salts [6], [22]. In this research, titanium dioxide (TiO₂) and titanium dioxide/graphene oxide (TiO₂/rGO) were prepared using the sol-gel method as referred to in previous research [4]. A titanium precursor solution of Titanium Isopropoxide (TTIP), ethanol, and acetic acid was stirred for two hours at 80°C (TTIP: acetic acid : ethanol 1:0.02:9 ratio). A GO solution was prepared by dispersing 40 mg of GO powder into 10 ml of ethanol. The solution was then ultrasonicated for 1 hour and was well dispersed in the solution. 0.1% of the GO solution was mixed in the TiO₂ solution for 1 hour at 80°C and rested at room temperature for a minimum of 8 hours before the spin coat. The obtained TiO₂/rGO was spin-coated on a Si/SiO₂ substrate and annealed in a tube furnace in the presence of H₂ and Ar gas at 600°C for 1 hour to crystallise the titanium dioxide and reduce the graphene oxide.

3.2. Fabrication of Electrodes

![Figure 1. Illustration of fabricated microelectrode devices with various gap sizes.](image)

The conventional photolithography process can produce stable and low contact resistances [23]. However, the use of chemicals during metallization or lift-off could compromise the unique properties of the materials. To overcome the limitations of the above techniques, the best approach to producing low resistance and stable electrical contacts is by using a shadow mask. A shadow mask is a precision, micromachined sheet or metal plate that is used in various metal deposition processes.

Si/SiO₂ substrate is used. Different microelectrode gaps (L₋) of 250 µm, 100 µm, and 50 µm were fabricated on the TiO₂ and TiO₂/rGO samples. Excellent uniformity of spin-coated TiO₂/rGO and TiO₂ was observed throughout the sample using Raman spectroscopy and FE-SEM. 80 nm-thick silver electrode was deposited on the sample using a stainless steel shadow mask, as shown in Figure 1 by thermal evaporation at a rate of 0.5 Å/s at 1.8 V and 1.1 Å/s at 1.3 V.

4. DATA COLLECTION AND ANALYSIS

4.1. Chemical Analysis by Raman Spectroscopy

Raman spectroscopy is a technique for non-destructive chemical examination that provides extensive information about chemical structure, phase and polymorph, crystallinity, and molecular interactions. It is based on light’s interaction with vibrated chemical bonds within a substance.

![Figure 2. Raman spectroscopy analysis for TiO₂/rGO and TiO₂ thin film.](image)
TiO$_2$ and TiO$_2$/rGO thin films were characterised by their Raman spectra, Figure 2, which allow for consideration of the conjugated, and carbon-carbon double bonds, which lead to high-intensity peaks in the Raman spectrum. The four typical Raman spectra of characteristic active modes of anatase TiO$_2$ with symmetries $E_g$, $B_{1g}$, $A_{1g}$, and $E_g$ were observed at 144, 400, 519, and 640 cm$^{-1}$ and peaks $B_{1g}$, $E_g$ and $A_{1g}$ for 140, 430, and 590 cm$^{-1}$, respectively, in rutile TiO$_2$ [24], [25] Figure 2 reveals the peaks of $E_g$, $B_{1g}$, $A_{1g}$, and $E_g$ at 139.06, 391.69, 515.12, and 632.76 cm$^{-1}$. These characteristics of vibrational frequencies and their intensity ratios confirmed the phase of pure anatase TiO$_2$. The peaks revealed by TiO$_2$/rGO area at 148.7, 397.48, 522.83, and 636.61 cm$^{-1}$, respectively. Moreover, the raman of TiO$_2$/rGO shows the D band peak at 1348.12 cm$^{-1}$, which is a characteristic peak that indicates the presence of defects, particularly in carbon-based materials like reduced graphene oxide.

4.2. Morphological Analysis by FESEM

Figure 3 displays an image of the surface morphology of TiO$_2$ and TiO$_2$/rGO thin films. The TiO$_2$ forms in anatase appear to consist of a few fused spheres, while TiO$_2$/rGO form imperfectly merged spherical shapes [26]. The surface of TiO$_2$/rGO was found to be relatively smoother when compared to anatase. This morphology is consistent with recent findings from previous investigations [22].

4.3. Electrical Characterization by Picoammeter

Figure 4 shows the current-voltage (I-V) measurement of the devices. The current output and resistance inside the channel were investigated using SMU instruments in DC mode with channel variations of 250 μm, 100 μm, and 50 μm. Both materials show that the smaller the microelectrode gaps, the lower the resistance of the channel, which results in less voltage drop across it for a given current.

![Figure 3. Morphological result of (a)TiO$_2$ and (b)TiO$_2$/rGO thin film on Si/SiO$_2$ substrates.](image-url)
Currently, the biosensor industry has focused on miniaturisation due to the possibility of achieving lower detection limits from micromolar to femtomolar. Previous studies prove that the most suitable output current for biosensor applications is in the range of microns and nanoampere [28]. The simulation study in previous work has shown that the TiO$_2$/rGO material increases its current performance by reducing the gap of the channel [21]. In this study, we have proved experimentally that TiO$_2$ and TiO$_2$/rGO could conduct in smaller channel lengths with current output in the range of microns.

6. CONCLUSION

We have developed microelectrodes using TiO$_2$ and TiO$_2$/rGO sol-gel spin-coating techniques. Raman spectroscopy and FESEM analysis have proven the presence and quality of both materials. I-V measurement shows that the microelectrode with the smallest gap size has the highest output current response for TiO$_2$ (8.04E-12 A) and TiO$_2$/rGO (2.385E-8 A) at 0.04V. In terms of electrical characterization, the two electrodes show that TiO$_2$/rGO has better conductivity than TiO$_2$. It is due to the presence of rGO. However, both materials show a good potential interface contact that can act as a great transducer material for biosensor applications. The straight-forward steps of synthesis material presented in this work will facilitate fabrication of biosensor devices in the future while at the same time lowering the cost of manufacturing and preventing thermal denaturation of bio-samples.

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