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Sensors for Polystyrene Nanoplastics Detection in Water Samples

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ABSTRACT

Polystyrene nanoplastics (PS-NPs) are increasingly discovered in aquatic ecosystems, posing ecological and human health problems. Identifying PS-NPs in complex environmental matrices remains difficult due to their small dimension, chemical passiveness, and the existence of background interferences. Recent improvements in sensor and biosensor technologies have shown promise for improving the sensitivity, selectivity and portability of PS-NPs detection systems. However, maintaining high detection accuracy while minimizing false signals and matrix interferences remains a significant challenge. Limited selectivity, sensor fouling, and a lack of standardization across detection platforms add to the complexity of analysis. To improve recognition performance, several ways have been investigated, including surface functionalization, nanomaterial immobilization and substrate modification. These strategies seek to overcome limitations in sensitivity, repeatability, and environmental applicability. As a result, more advanced sensor platforms capable of detecting low concentrations in real time are urgently needed. This study highlights recent research on sensor and biosensor technologies used to detect PS-NPs, with a focus on the integration of nanomaterials and molecular recognition elements. Literature from ScienceDirect and IEEE Xplore indexed journals until June 2025 was reviewed. The findings indicate that hybrid sensor systems, particularly those employing plasmonic nanoparticles and functionalized nanostructures, are at the forefront of PS-NPs detection research. To find novel sensing methodologies, this review examines current technologies and compares their merits using recent case studies and performance analysis.

Keywords: Sensor, Biosensor, Polystyrene, Nanoplastics

1. INTRODUCTION

Since the first synthetic plastic was produced in the early 20th century, the worldwide population has become increasingly dependent on plastic. Plastics are highly beneficial in low production costs, stability, light weight, and adaptability, which has increased demand worldwide. With an annual rapid growth rate of 15% over the last 11 years [1], Malaysia alone is home to 1300 plastic manufacturing factories, making it one of the world's major plastic businesses [2]. Unfortunately, this prolific production activity has contributed to the rising amount of plastic waste. According to statistics, Malaysia was the eighth-largest plastic polluter in the world in 2021 and over 0.94 million tons of improperly managed plastic waste are produced annually in the country [2, 3].

Polystyrene (PS) is one of the commonly used thermoplastics, as it is versatile. It is also regarded for its low cost and broad application in packaging, insulation and disposable materials [4]. Unfortunately, extended exposure to styrene is associated with neurotoxicity that results in

headaches, light-headedness, and possible cancerous effects. Hence, IARC has categorized polystyrene as a potentially carcinogenic substance (carcinogenicity class B2) [5].

Initially, PS plastic waste came from land disposal, wastewater treatment, tire wear, paint failure, textile washing, and marine casualties. These PS plastic waste are then accumulated, sank, and are consumed by creatures and re-distributed by currents in surface water [2, 6]. After going through significant environmental degradation, the plastics are broken into smaller pieces. This degradation results in the formation of polystyrene microplastics (PS-MPs) and [7–9] polystyrene nanoplastics (PS-NPs). These PS-MPs and PS-NPs can be categorized as primary or secondary sources. The industry produces primary microplastics, including PS-MPs, which are then released into the environment [10]. Particles are primarily sourced from cosmetics, medicines, paints, medical equipment, and electronics. In the cosmetic sector, they are usually components of face cleaning products, hand washing exfoliants, and peeling formulae [11]. NPs are also released

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during high-temperature engineering procedures. Other significant sources include plastic packaging manufacturing factories. Secondary sources are plastic particles that disintegrate into smaller pieces due to physical and chemical processes [12]. Improper plastic waste handling leads to the discharge of particles rinsed from synthetic fibers (**Figure 1**).

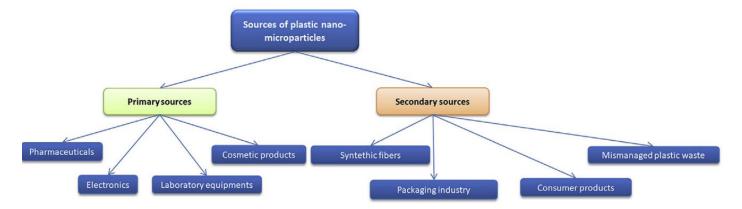


Figure 1. Primary and secondary sources of micro- and nanoplastic in the environment [5].

PS-NPs travel in a variety of ways, including air deposition, wastewater discharge, and disintegration from bigger plastics. PS-NPs' compact size allows them to reach even the most isolated habitats, from deep oceans to mountain peaks. PS-NPs frequently vary in form and surface structure, with diameters ranging from 1 to 100 nm [13]. Small particles have a higher surface area-to-volume ratio than bigger particles, which boosts their reactivity and potential for interaction with adjacent molecules. PS-NPs prefer to clump together in aquatic environments, which influences

how they settle and their availability in marine habitats [14]. Additionally, PS-NPs can carry other pollutants, alter their environmental impact and potentially make them more toxic [15]. Studies have also verified that PS-NPs penetrate living organisms from the aquatic environment and accumulate in subsequent links of the food chain. It is found that PS-NPs enter embryos and accumulate in the yolk sac [16]. Another study by Mattsson *et al.* (2015) found accumulation of 24 and 27 nm PS-NPs in the food chain [17]. This accumulation is summarized in **Figure 2**.

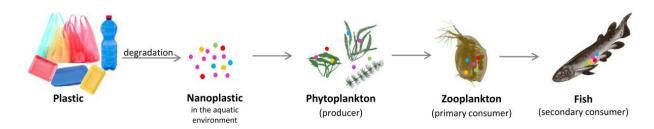


Figure 2. Accumulation of polystyrene nanoparticles in the food chain [5].

Due to their tiny size, widespread distribution and ability to accumulate, PS-NPs are extremely dangerous to living things in aquatic and marine environments. It has been demonstrated that exposure to 44 nm PS-NPs alters the bioenergetics of developing zebrafish, impacting their growth and metabolic functions. At the cellular level, PS-NPs also raise the zebrafish's cardiac apoptosis and reduce mitochondrial efficiency in ATP synthesis [18–24].

It has also been demonstrated that PS-NPs change gene expression and upset reproduction processes. A long-term exposure of water fleas *D. pulex* and *D. magna* to PS-NPs concentrations of 50 and 75 nm changes the gene expression implicated in stress defence and heat shock proteins, which impacts the antioxidant system and damages the organism [25–27]. After 3 weeks of PS-NPs exposure, the entire transcriptome of *D. pulex* was tested, revealing changes in the genes related to trehalose transport and metabolism, chitin metabolism, growth-related genetics, protective mechanisms, and sex

differentiation. These changes resulted in ageing, delayed development and fertilization, and modifications to the population's sex and reproductive structure.

Despite their expanding environmental impact, detecting PS-NPs remains technically difficult. This is due to their nanoscale size, chemical complexity, and prone to congregate in aquatic environments [28]. Conventional techniques, such as electron microscopy and dynamic light scattering, provide excellent sensitivity but are constrained. Some barriers are included by the high cost, extensive sample preparation, and low output [28, 29]. As a result, sensors such as electrochemical, optical, and biosensing platforms have gained popularity. These sensors are known for their ability to provide real-time, portable, and low-cost alternatives.

Furthermore, advancements in π - π interaction-based recognition, functionalised nanostructures, and hybrid sensing systems provide exciting opportunities. As a result,

this study presents a systematic literature review (SLR) utilising the PRISMA technique, with a focus on existing sensors for PS-NP detection in water samples. Section 2 describes the review methods. Section 3 summarizes and compares sensors for PS-NPs detection, including their performance metrics. Finally, in section 4, the conclusion is made.

2. REVIEW METHOD

The fundamental idea of a sensor in detecting PS-NPs is based on physicochemical properties that are altered when PS-NPs interact with the sensing element in general. The method enables PS-NPs detection and quantitation by converting its chemical signal of chemical compound concentration into a measurable change in the sensor's response [30, 31]. As polystyrene itself is a synthetic polymer, many studies are focused on the modification of the sensing surface or various sensing element approaches. This includes integrating the substrate with different types of receptors or bioreceptors such as nanostructure, protein, natural or synthetic polymer, aptamer, antibodies, enzymes and dyes [32–34].

Linear range, sensitivity, limit of detection (LoD) and relative standard deviation (RSD) are important indicators in evaluating a sensor's performance in detecting PS-NPs. Linear range refers to the concentration interval in which the signal response is linearly proportional to analyte concentration [35]. Sensitivity is the slope of that response

[35, 36]. The LoD represents a minimal detectable concentration that can be differentiated from noise [36]. RSD measures repeatability and reliability over several measurements [37]. These factors collectively provide a full assessment of sensor performance in detecting PS-NPs [38, 39]. This research presents a thorough analysis of current methods for detecting PS-NPs in water samples. Advanced evaluation is one of the most important issues right now. Thus, a systematic flow strategy was adopted in this study. A systematic review is a way for deliberately discovering patterns, trends, and important assessments of the literature on study themes. The review technique included four steps to choose relevant papers for the study. This study utilized the Preferred Reporting Items for Systematic Reviews and Meta-Analyses (PRISMA) approach [40]. This approach depicts the flow of information across the stages of a systematic review, as shown in **Figure 3**.

The first stage in producing a literature review is to locate relevant research items to the research question. The total number of papers searched was screened. Papers were evaluated for eligibility based on their abstracts. Further research was recommended to address the problems raised. This study employed the best practice method to perform a complete literature review. The publishing rules gave valuable information to help researchers evaluate the accuracy of the review. Furthermore, an inquiry into the systematic analysis of the many papers included in this review. The approaches were studied using the ScienceDirect and IEEE Xplore databases.

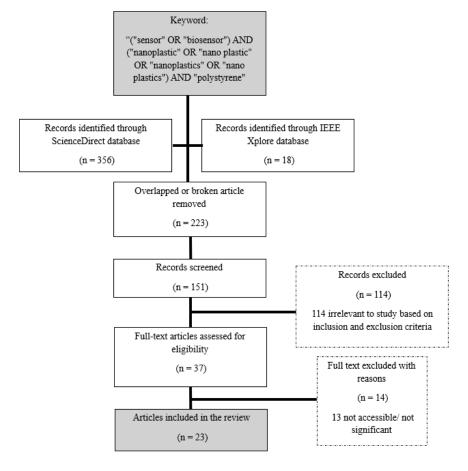


Figure 3. PRISMA flow diagram of the entire procedure in selecting reviewed articles.

2.1. Preliminary Identification

A selection of studies on different sensors in detecting PS-NPs was found in the literature. Keywords like "polystyrene", "nanoplastic," "sensor," and "biosensor." To consider all relevant recent studies, the year was limited to 2021. ScienceDirect and IEEE Xplore databases were used to search the literature and articles. As shown in **Table 1**, the first search produced 356 articles from the ScienceDirect database and 18 articles from the IEEE Xplore database.

2.2. Screening

Screening identifies suitable research articles that correspond to predefined research questions. The first

screening rejected 223 publications due to overlapped or visible broken articles, and the remaining 151 papers were examined using the study's exclusion and inclusion criteria as described in **Table 2**. The first need was to include research publications that provided practical guidance, such as reviews, encyclopedia, index, book series, chapters, and conference proceedings. These papers have been eliminated from recent investigations. English-language publications issued in 2025 are discussed and analysed. 114 articles were eliminated due to premature results and a lack of debate regarding the detection of PS-NPs in water samples. Some articles were partial or unavailable, with broken links and overlaps.

Table 1 Search strings from the ScienceDirect and IEEE Xplore databases

Database	Search Strings	Results	
ScienceDirect	("sensor" OR "biosensor") AND ("nanoplastic" OR "nano plastic" OR "nanoplastics" OR "nano plastics") AND "polystyrene" Refined by: Document types: (Article) and Publication Year: 2025	356 Articles	
IEEE Xplore			

 Table 2 Second-stage examination criteria

Criterion Inclusion		Exclusion
Language	English	Non-English
Published Year	2025	<2025
Source Type	Journal (only research articles)	Conference proceeding
Document Type	Article	Letter, review, conference, and note
Research Area	Environmental and Engineering	Besides Environmental and Engineering

2.3. Eligibility

After meeting all inclusion and exclusion criteria, the final review sample was generated. The research objectives required articles that addressed studies and research in sensor development for detecting PS-NPs. 14 publications were excluded as their titles and abstracts did not align with the study's purpose based on empirical data. Finally, 23 papers and articles were made available for evaluation, as depicted in **Figure 3**.

3. PERFORMANCE METRICS OF CURRENT SENSORS FOR PS-NPS DETECTION

As the impacts of plastic pollution have become an alarming concern globally, many scientists and researchers are developing the best sensor to detect PS-NPs at the lowest concentration. Additionally, these studies are primarily focused on aquatic and marine habitats in environmental monitoring and analytical studies. However, precisely detecting and measuring PS-NPs remains difficult due to their nano-scale size, low ambient concentrations, and the complexity of sample matrices. In response, a comparative study of modern sensor and biosensor technologies was done to highlight the importance of material selection, surface functionalization, substrate immobilization and

signal amplification methodologies. The performance metrics of these sensors were evaluated using linear range, sensitivity, limit of detection (LoD) and relative standard deviation (RSD). This review looked at how new sensing techniques, such as optical, electrochemical, and receptor-based platforms, can increase the efficiency and reliability of detecting PS-NPs in water samples. Researchers have devised a variety of approaches to overcome constraints in selectivity, interference, and detection in real-world samples. Notably, nanomaterial-enhanced sensors, such as those containing metal nanoparticles and protein, have

received interest due to their high sensitivity and customizable surface chemistry. A rigorous literature search resulted in the selection of 23 peer-reviewed studies for extensive analysis and comparisons.

Many studies and experiments are being conducted to detect the lowest concentration possible of PS-NPs. Hence, **Table 3** summarizes and tabulates the recent sensors development and design, their linear range, sensitivity, LoD, RSD performance metrics and their advantages or novel approaches from the studies.

Table 3 Summary of recent sensors for PS-NPs detection

References	Sensors platform	Modification or material used	Linear Range (µg/mL)	Sensitivity	Limit of Detection (LoD) (g/mL)	Relative Standard Deviation (RSD) (%)	Advantages / novel approaches
[43]	Electrochemical	Silver (Ag) labelling	50 – 600	4.604 μΑ/(mg/mL)	13 μ	6.03 – 10.59	Simple. anti- interference, and low-cost.
[44]		Polyaniline coating magnetic biochar nanoparticles (PA@MBC _{BM})	0.03 - 1	2.15 μΑ/(μg/L)	1.26 n	5.4 – 13.3	Outstanding elimination was assured in both cold and warm water with a pH range of 1 to 10.
[45]	Photoelectroche mical- electrochemical	CdS/CeO ₂ heterojunction	0.5 - 800	7.92 μΑ/(μg/L)	0.38 n	0.4 - 2.97	The sensor displays selected modes and self- checking functions to ensure accurate detection
[46]	SERS	Aluminium and copper as nanoparticle-on- film (NPoF)	0 - 1000	0.92 a.u./(μg/mL)	1 μ	-	Plasmonic coupling amidst third- generation hotspots (TGHs) and second- generation hotspots (SGHs)
[47]		Nickel Oxide (NiO) nanosheet array decorated with Silver Nanoparticles (AgNPs)	100 – 2500	Drinking water: 1.77 a.u./(μg/mL) Milk: 0.396 a.u./(μg/mL) Liquor: 0.889 a.u./(μg/mL)	5 μ	4.70 - 8.54	Unique nanowell architecture alongside silver plasmonic enhancement
[48]		Honeycomb-like AgNPs@TiO2 array	100 - 2500	Tap water: 1.14 a.u./(μg/mL) Lake water: 1.34 a.u./(μg/mL)	100 μ	0.49 – 10.23	Novel honeycomb- like AgNPs

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				Soil: 1.44 a.u./(μg/mL) Sea water: 0.51 a.u./(μg/mL)			
[49]		Silver nanoparticles (AgNPs) anchored on Nb ₂ CT _x (MXene)	0.1 – 100	131 a.u/(μg/mL)	100 n	2.88 - 8.30	Exhibits excellent stability of PS- NPs on different detection sites
[50]		Mesoporous spiked gold nanocrystals (AuNCs) are homogeneously deposited on a glass fiber filter membrane	0.1 - 200	-	100 n	6.91 - 7.62	Combination of membrane filtration technology and SERS
[51]		three-dimensional Ti ₃ C ₂ T _x /TiO ₂ /W ₁₈ O ⁴⁹ semiconductor heterostructures	25 - 1000	Rainwater: 0.7804 a.u./(µg/mL) Soil: 0.8120 a.u./ µg/mL) Industrial wastewater: 0.6033 a.u./(µg/mL)	25 μ	12.58	Sensor's fingerprinting features enable the identification of other plastic types including polyethylene (PE) and polyethylene terephthalate (PET)
[52]		Polydomain aggregating silver nanoparticles (PASN) on a hydrophilic and a punched hydrophobic PVDF combined filter membrane	1 - 500	Tap water: 1360.03 a.u./ μg/mL) Lake water: 1677.53 a.u./ μg/mL)	1 μ	8.6	Using silver nanoparticles of different sizes to increase hotspot areas
[53]		Metal-free Cu₂S@MoS₂ nanoarray	50 - 2500	0.5 a.u./ μg/mL)	50 μ	10.3	Special approach of metal-free Cu2S@MoS2 nanoarray provides great sensitivity and repeatability
[54]		Gold sputtered glass slide cover	10 - 40	-	0.32 μ1	-	Experiment was carried out in the presence of interferences such as salts, sugars and amino acids to demonstrate great selectivity
[55]		Gold nanoparticles (AuNPs)	800 - 10,000	-	800 μ	-	Employing electric charge adsorption to

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							enhance PS- NPs' SERS detection signal.
[56]		Nanowall SERS heterocomposite based on 3D D(C ₇ CO)-BTBT nanowalls and 2D few-layer MoS ₂ flakes	0.4 - 50	-	0.45 μ	-	SERS property was enhanced by the synergistic effects of electromagneti c and charge transfer
[57]	Fluorescence	Organic dyes; Rhodmine6G, coumarin521, and fluorescein	-	-	20 n	-	Rapid analytical separation of dyes by capillary electrophoresis
[58]		Fluorescence Lifetime Analysis (FLA)	10 - 1000	-	10 μ	-	Endogenous fluorescence has been used to identify PS- NPs materials.
[59]	Optical	metal-organic framework (MOF) ZIF-8 functionalized S- tapered fiber (STF)	100 - 800	114.1353 nm/% (w/v)	1.8 μ	-	Strong electrostatic adsorption effect and π-π stacking with PS-NPs
[60]		Microfiber	1 × 10 ⁻⁸ – 10	1.229 nm/log10(mg/mL)	1.23 n	2.443	Operando on- line quantification of PS-NPs
[61]	Optofluidic	Laser- backscattered fiber-embedded optofluidic chip (LFOC)	0 - 100	0.00273 /(μg/mL)	0.23 μ	0.50 - 6.36	LFOC unique features include plug and play, reagent-free, field- deployment, low-cost
[62]	Colorimetric	single-atom nanozyme of zeolitic imidazolate framework (ZIF- FeSAN) was prepared using hemoglobin as template and Fe- source	0 - 20	8.1 a.u./(mg/m L))	212 n	0.61 - 4.48	ZIF-FeSAN possess good adsorption characteristics for PS-NPs via electrostatic and π-π interactions
[63]		Mercaptoundecano ic acid (MUA)- functionalized gold nanoparticles (AuNPs)	0 - 1300	0.616/(mg/ mL)	26 n	10	Able to quickly visualize the separation of PS-NPs less than 100 nm
[64]	Triboelectric nanogenerators (TENG)	BaTiO ₃ -PDMS nanocomposite films	0 - 0.08	9.44 V/μwt%	0.823 n	0.084 - 1.13	Combined with Deep Learning algorithm to identify PS-NPs
[65]	Spectroscopy	Mie scattering	0.1 – 10	139.5 a.u./ (g/mL)	4.2 n	3.8 - 5.4	This method instantly detects a sample within

			20 seconds, removing the requirement for
			pretreatment.

3.1. Linear Range and Sensitivity

Linear range is the concentration window within which a sensor maintains proportionality between signal and analyte concentrations. As for sensitivity, it quantifies signal change per unit analyte. If a sensor responds linearly, its sensitivity remains constant across its range and equals the slope of the straight-line plot [35]. Although these parameters are closely related, they are not directly proportional. In Table 3, SERS sensors possessed extremely high sensitivity but have narrower linear ranges due to signal saturation. However, some of the SERS sensors contain wide linear ranges but have lower sensitivity to maintain proportionality across diverse concentrations. AgNPs@TiO2 array SERS sensor by Li et al. (2024) has a small working window skewed towards high concentrations, at 100 to 2500 μ g/mL) [48]. This sensor is suitable for higher concentration samples that require substantial plasmonic amplification. Mesoporous Au nanocrystals SERS sensor gives a greater dynamic range at 10 to 1000 μg/mL [50]. Polydomain aggregating silver nanoparticles (PASN) SERS sensor has the steepest calibration slope, up to 1677.53 a.u./μg·mL⁻¹. This indicates substantial signal enhancement and is ideal for exact quantification and detection of PS-NPs at moderateto-high concentrations [49, 52]. TiO₂ semiconductor heterostructures SERS sensor, however, are less susceptible to changes in concentration, indicated by their slightly different sensitivity values when tested on various water samples [51].

The microfibre optical sensor has a wide linear range of 1 \times 10⁻⁸ µg/mL to 10 µg/mL. Hence, making it ideal for monitoring PS-NPs at ultra-trace and ambient levels [60, 66, 67]. The LFOC optofluidic sensor has poorer sensitivity at 0.00273 /µg·mL. This indicates a slower reaction to incremental concentration changes [67]. However, their lower sensitivity is frequently offset by their mobility, realtime monitoring capability, and reagent-free operation. Both colorimetric sensors fall somewhere in the centre, offering modest sensitivity while remaining user-friendly and low-cost. MUA-functionalized AuNPs colorimetric sensor by Zhou et al. (2023) offers a wider linear range interval [63]. However single-atom nanozyme of ZIF-FeSAN colorimetric sensor by Hu et al. (2024) offers better sensitivity [62]. Hence, both sensors are fit for quick screening but less effective for extremely precise detection in real water samples. This variance in linear range demonstrates a compromise between sensitivity augmentation techniques and capacity to retain linearity over a wide concentration range [66, 67].

3.2. Limit of Detection (LoD)

Based on Table 3, Surface-enhanced Raman spectroscopy (SERS) sensors are the most developed and explored approach in detecting PS-NPs. SERS are integrated or modified with different nanomaterials to enhance their sensitivity and selectivity. AgNPs anchored on Nb₂CT_x (MXene) SERS sensor [49] and Mesoporous spiked AuNCs SERS sensor [50] both have the lowest LoD compared to other SERS approaches at 100 ng/mL. As polystyrene uniquely possesses aromatic structure of benzene rings, this structure creates distinct Raman peaks, such as those at $\sim\!1000~\rm cm^{-1}$ and $\sim\!1600~\rm cm^{-1}$ [46, 55]. Its distinctive vibrational modes make it easily detectable in SERS spectra, even at low concentrations.

Other sensor platforms such as polyaniline coating magnetic biochar nanoparticles (PA@MBCBM) electrochemical sensor, organic dyes fluorescence sensor, microfiber optical sensor and MUA functionalized AuNPs colorimetric sensor have the lowest LoD at 1.26 ng/mL [44], 20 ng/mL [57], 1.23 ng/mL [60] and 26 ng/mL [63], respectively. However, the lowest LoD is captured by CdS/ CeO₂ heterojunction photoelectrochemical-electrochemical sensor by Xiao et al. (2024) at 0.38 ng/mL [45]. CdS is a semiconductor that absorbs light well yet has a high electron-hole recombination rate. Combining it with CeO2, which has stable redox characteristics (Ce^{3+}/Ce^{4+}), results in a heterojunction that promotes charge separation and photoactivity [68, 69]. CdS/CeO₂ is a highly appealing substance for dual-mode sensors, particularly in addressing electron transfer difficulties induced by protein interactions with PS-NPs in aquatic settings.

Some reagent package inserts may state that an assay has a dynamic range that extends from zero concentration to an upper limit. However, most assays cannot properly quantify analyte concentrations at zero. A minimum concentration of analyte is required to provide an analytical signal that can be consistently separated from background noise. This reflects the signal observed in the absence of the analyte [70]. Hence, several approaches are used in some of the studies reviewed in this paper to estimate the LoD. This includes visual definitions, signal-to-noise ratio estimation (three or two times the noise level), calculation from the standard deviation of the blank and calculation from the calibration line at low concentrations [36, 71].

Polyaniline coating magnetic biochar nanoparticles (PA@MBC_{BM}) electrochemical sensor by Wang *et al.* (2025) calculated LoD using the $3\sigma/S$ formula, where σ and S represent blank standard deviation and slope of the linear curve, respectively. With LoD at 1.26 ng/mL, it is lower than nanoplastics electrochemically labelled with Ag, at 13 μ g/mL. Similarly, this Ag labelling electrochemical sensor developed by Zhao *et al.* (2024) also applied the 3N/S

formula to calculate LoD. The N represents the standard deviation of the blank signal and S represents the slope of the calibration curve [43, 72]. This same formula is also applied in calculating LoD for LFOC optofluidic sensor [61] and single-atom nanozyme ZIF-FeSAN colorimetric sensor [62].

CdS/CeO₂ heterojunction photoelectrochemicalelectrochemical sensor by Xiao et al. (2024) calculated LoD by using $3 s_b/m$ formula, where s_b is the standard deviation of the response and m is the slope of the calibration curve [45, 73]. Gold sputtered glass slide cover SERS sensor by Chaisrikhwun et al. (2023) calculated LoD based on the standard deviation of the response (σ) and the slope of the calibration curve (S), using the equation, LOD = $3.3\sigma/S$ [54]. BaTiO₃-PDMS nanocomposite films TENG sensor calculated LoD by using the formula LoD = $K\sigma/k$. Where K is the confidence factor, generally taken as 3, σ is the standard deviation of the blank sample, calculated to be 0.2593, and k is the slope of the standard curve over the low concentration range [64].

3.3. Relative Standard Deviation (RSD)

RSD is an important performance parameter for sensors since it represents the accuracy of repeated measurements. A low RSD (e.g., < 5%) value implies that the sensor itself can deliver consistent findings under the same settings, hence proving repeatability [74]. Maintaining minimal RSD values is necessary to ensure reproducibility across multiple runs in laboratories and real sample settings [49]. When comparing across the sensor platforms, both the microfiber colorimetric sensor and the LFOC optofluidic sensor have continuously low RSD values. Both sensors have RSD values of less than 5%, which makes them the most reproducible platforms [60, 61]. For the SERS sensor, it varies in sensitivity and material design. The honeycomblike AgNPs@TiO2 array SERS sensor has the lowest minimum RSD at 0.49%, indicating exceptional precision under optimal conditions. However, its wider range of up 10.23% indicates more variability measurements. AgNPs anchored on Nb₂CT_x (MXene) SERS sensor have a narrower and more stable RSD range (2.88-8.30%), indicating improved consistency and reliability. Although the honeycomb-like AgNPs@TiO2 array SERS sensor can achieve higher repeatability in the best scenario, the MXene-based SERS sensor has lower total RSD variability, making it more trustworthy throughout repeated tests.

Based on Table 3, the BaTiO $_3$ -PDMS nanocomposite films TENG sensor has the lowest RSD (0.084 to 1.13%) as compared to other sensors. This unusually low RSD indicates that TENG sensors are less susceptible to signal fluctuations and external disturbances. The sensor's high repeatability also boosts credibility in controlled laboratory tests and real samples, where sample heterogeneity and operational unpredictability can challenge sensor performance.

3.4. Nanomaterials

Metal nanoparticles are one of the most utilized nanomaterials to increase the device's performance in terms of analyte recognition. Metal nanoparticles propose unique features such as optical, electrical and excellent catalytic properties [33, 75–77]. SERS performance was connected to the number of PS-NPs surrounded by the gold nanoparticles (AuNPs), and the strength of the superposition electric field would have a gradual linear relationship with the number of charged surrounding PS-NPs, as proven by Lan *et al.* (2024) [55].

Some nanomaterials, particularly noble metals such as AgNPs, raise concerns about their environmental safety. While AgNPs contribute significantly to sensitivity enhancement through localized surface plasmon resonance, as demonstrated in NiO/AgNPs nanowall SERS sensor by Li et al., 2025 [47] and AgNPs anchored on Nb₂CT_x MXene SERS sensor by Chang et al. (2024) [50], respectively. They are also known to be toxic to aquatic organisms when released in excess. The toxicity mechanism of AgNPs is mostly owing to their ability to create a huge amount of Reactive Oxygen Species (ROS), which can damage the cell membrane and cause cell death. Specifically, AgNPs are quickly oxidized by O₂ and other molecules, resulting in Ag+. Excessive and continuous ROS will cause oxidative stress [78, 79]. Leaching of free Ag+ ions or unbound nanoparticles into the environment might lead to secondary pollution, worsening the conditions the sensors are monitoring. Because of the dual nature of AgNPs, more research is needed into their long-term environmental impact, biocompatibility, and potential leaching behavior during sensor deployment in the

Interestingly, some sensor materials can not only detect PS-NPs but also reduce their toxicity by their anti-microbial properties. Wang et al. (2025) produced a polyanilinecoated magnetic biochar (PA@MBCBM) electrochemical sensor, which was demonstrated to be effective throughout a wide pH and temperature range and contributed to the elimination of PS-NPs during detection [44]. This dual functionality provides a viable avenue for developing a multifunctional sensing device that integrates detection and remediation. Similarly, the ZIF-FeSAN nanozyme colorimetric sensor revealed high adsorption of PS-NPs via electrostatic and π - π interactions by Hu et al. (2024), indicating a potential "trapping" effect that may limit the bioavailability of nanoplastics in water [59]. These findings highlight a new trend in sensor research: developing materials that not only detect contaminants with great precision but also contribute to their neutralization or sequestration.

3.5. Sensors interaction with PS-NPs

Aromatic π - π interactions are a crucial chemical recognition technique in developing sensors for PS-NPs recognition due to the strong affinity between the aromatic rings of polystyrene and the π -conjugated sensor materials [80]. In an electrochemical sensor with polyaniline coating, magnetic biochar nanoparticles (PA@MBCBM)

electrochemical sensor by Wang *et al.* (2025), PS-NPs adsorption resulted in enhanced binding energy and aromatic carbon content (C=C, sp²), indicating that π - π interaction was responsible for their attachment [44].

Xiong *et al.* (2024) discovered that ZIF-8 possesses a high electrostatic adsorption interaction with PS-NPs. The benzene-like ring structure of ZIF-8 and PSNPs makes π - π stacking more effective for adsorption. ZIF-8 serves as a bridge, communicating changes in PS-NP concentration and evanescent waves across the S-tapered fibre (STF) surface [59]. Hence, a high-sensitivity optical-fiber sensor is achieved. Similar findings are also demonstrated by Li *et al.* (2024), the fabricated microfiber optical sensor can continuously detect PS-NPs via strong benzene ring π - π stacking interaction between the L-Phenylalanine (L-PHA) and the PS-NPs particles (**Figure 4**) [60]. A single-atom

nanozyme of zeolitic imidazolate framework (ZIF-FeSAN) colorimetric sensor developed by Hu et~al. (2024) revealed that ZIF-FeSAN has significant adsorption for PS-NPs with the active centre of Fe atom shielding, and peroxidase-like activity can be greatly lowered after adsorption of PS-NPs. This is due to electrostatic and π - π interaction [62].

The Ag labelling electrochemical sensor developed by Zhao *et al.* (2024) utilized the electrochemical reaction of Ag to quantify PS-NPs. Electrostatic interaction connected Ag+ to the PS surface. The Ag+ surface was then reduced to Ag nanoparticles (AgNPs) by NaBH4. This allows the AgNPs to interact with PS-NPs, resulting in an Ag-PS complex. The free AgNPs were isolated from the Ag-PS using standing flotation. Hence, the concentration of PS-NPs was calculated by measuring the current response of Ag [43].

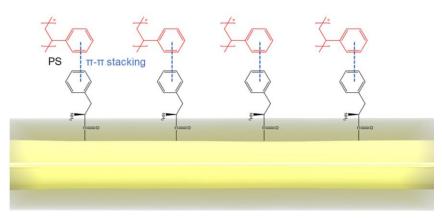


Figure 4. Schematic illustrations of the binding of L-PHA probe molecules with PS-NPs on the surface of optical microfiber [60].

3.6. Other Performance Metrics

High coefficients of determination (R²) and high recovery rate sensor peaks demonstrate consistent hotspot distribution and signal intensity throughout several tests. High R² is necessary in environmental monitoring to guarantee field measurement confidence, particularly when detecting extremely low PS-NPs concentrations, where even small errors might cause contamination levels to be misinterpreted. Yang et al. (2025) portable SERS chessboard substrate displays excellent coefficients of determination with 0.9916 or 99.16% of R^2 [52]. Furthermore, the use of a single-atom nanozyme-enabled strategy for rapid, visual, and real-time detection, as demonstrated by Hu et al. (2024). The sensor offers high satisfactory recoveries, ranging from 91.47% to 108.12% and good selectivity, which makes it useful for different water samples [62].

Spectroscopy sensors with implying principles of Mie scattering have good anti-interference ability. The approach was confirmed by the sensor's satisfactory recoveries, ranging from 82.4% to 110.3% of the self-ground PS-NPs. The approach directly measured the total amounts of NPs in the five commercial bottled water brands [65]. Satisfactory recoveries indicate the sensor provides accurate and reliable measurements of the target analyte with minimal error. These findings verified that optimizing the sensor design and incorporating advanced

nanomaterials can result in excellent repeatability, which is crucial for environmental applications requiring reliable data over long-term deployments.

CONCLUSION

In conclusion, recent advances in sensor technologies for detecting PS-NPs have resulted in significant increases in sensitivity, selectivity, and repeatability, with several platforms now capable of detecting PS-NPs at environmentally relevant levels. Many sensors utilized nanomaterials to enhance the ability to identify the target analyte. Nanomaterials such as AgNPs in sophisticated heterostructures have been crucial in improving signal output and detecting precision. However, concerns surround the environmental toxicity of some nanomaterials, particularly AgNPs. This danger of nanoparticle leaking remains a major issue that deserve additional exploration. However, many studies have discovered that the AgNPs synthesization methods produced by the green method are safer and have higher biocompatibility when exposed to marine and aquatic habitats. Many studies also have discovered that some nanomaterials offer anti-microbial properties in reducing AgNPs toxicity. This can benefit in a variety of ways, including lowering PS-NP accumulation in marine life and organisms, limiting the formation of damaging biofilms, and directly counteracting PS-NP toxicity on cells and tissues. Additionally, π - π interactions play a vital role in PS-NPs sensor design, as demonstrated across various studies of π -conjugated sensor materials, enabling strong adsorption and high sensitivity through $\pi - \pi$ stacking with the benzene rings of PS-NPs. Several sensor platforms also promise dual functionality, allowing for both detection and partial remediation of PS-NPs in water systems. Moving forward, next-generation sensors should focus on increasing sensitivity to increasingly tiny plastic contaminants, such as pico-sized plastics, while also combining biocompatible and environmentally acceptable materials. Future designs should also focus on sensor miniaturization, real-time in situ detection capability, and interaction with the Internet of Things (IoT) for environmental monitoring and risk assessment.

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